

Synthesis and Upconversion Properties of Er^{3+} – Yb^{3+} co-doped LiBaBO_3 phosphor

S.P. Hargunani

Department of Physics, G.S. College, Khamgaon, Buldhana (M.S.), India

Abstract: Lithium barium borate phosphor (LiBaBO_3) doped with rare earth elements $_{0.02}\text{Er}^{3+}$ and $_{0.08}\text{Yb}^{3+}$ has been synthesized by solution combustion technique. For the first time Upconversion phenomenon was investigated in the samples at fixed concentration of rare earth dopants Er^{3+} and Yb^{3+} . Study shows that the synthesized materials emits in visible region after excitation in Infrared region. $\text{LiBaBO}_3:_{0.02}\text{Er}^{3+},_{0.08}\text{Yb}^{3+}$ shows the absorption in infrared region i.e. 820 nm to 1080 nm and at 545nm, 656nm respectively. At the excitation of 980 nm it shows the emission peakings at 590 nm and 596 nm. Upconversion mechanism was investigated in detail and attributed to the efficient resonant energy transfer from Yb^{3+} to Er^{3+} ions in this host material. Our results suggest a potential borate phosphor for Natrium Yellow and Amber upconversion phosphor.

Keywords: Lithium barium borate phosphor; Upconversion; Rare earth dopants; Solution combustion synthesis.

1. INTRODUCTION

Upconversion phenomenon is now becomes an interesting topic of research due to wide range of applications i.e. from medical [ⁱ] to solar energy sector. Spectral mismatch losses in solar cell were reduced by upconversion materials[ⁱⁱ]. Upconversion phosphors are used as a source of white light. Upconversion materials absorb near infrared light and re-emit in red, green, and blue. This three colors are combine to create a white-light source [ⁱⁱⁱ]. Recently, phosphors based on borates have attracted much attention due to their high stability, low synthetic temperatures, and high ultraviolet and optical damage threshold [^{iv}, ^v, ^{vi}]. Borates have been used as optical materials for second harmonic generation or mostly materials for fluorescence. Borate crystals are intrinsically luminescent and show thermo luminescence and other interesting optical properties [^{vii}, ^{viii}].

In 1997 “Chr. Wyss et al used Yb^{3+} as co-dopant (sensitizer) in Er^{3+} doped (activator) laser hosts. According to them Er^{3+} has narrow absorption band from 970 to 1010 nm wavelength and Yb^{3+} has a broad absorption band from 900 nm to 1025 nm wavelength and a higher absorption cross-section.[^{ix}]. Recently in 2011 Subrata Das, et al reported Strong green upconversion emission from Er^{3+} – Yb^{3+} co-doped KCaBO_3 phosphor. Single monoclinic phase doped KCaBO_3 has capacity of large concentration of rare earth doping. It gives intense green emission ($^2\text{H}_{11/2}, ^4\text{S}_{3/2}$ to $^4\text{I}_{15/2}$) through cooperative upconverted emission on 980 nm excitation [^x].

The rare-earth ions Er^{3+} and Yb^{3+} are up-conversion activator and sensitizer ions, which emit green to red light effectively in different host lattices. Er^{3+} has a simple energy-level structure that consists of $^4\text{I}_{15/2}$, $^4\text{I}_{11/2}$, $^4\text{I}_{13/2}$, $^4\text{F}_{9/2}$, $^4\text{S}_{3/2}$, $^2\text{H}_{11/2}$, and $^4\text{F}_{7/2}$ states [Figure (1)]. Usually, the $^2\text{H}_{11/2}$, $^4\text{S}_{3/2}$ to $^4\text{I}_{15/2}$ and $^4\text{F}_{9/2}$ to $^4\text{I}_{15/2}$ transitions lead to green and red emissions, respectively, in Er^{3+} -activated

phosphors. The luminescent properties of materials activated by Er^{3+} ions can be enhanced by the addition of Yb^{3+} ions via resonant energy transfer from Yb^{3+} to Er^{3+} through the absorption of a 980 nm photon [^{xi}].

In this work the upconversion emission properties of LiBaBO_3 co-doped with Er^{3+} and Yb^{3+} ions was systematically investigated for the first time. We report the synthesis of single monoclinic phase doped $\text{LiBaBO}_3:_{0.02}\text{Er}^{3+},_{0.08}\text{Yb}^{3+}$. It gives Natrium Yellow and Amber emission ($^4\text{F}_{9/2}, ^4\text{S}_{3/2} \rightarrow ^4\text{I}_{15/2}$ in the Er^{3+}) through cooperative upconverted emission on 980 nm excitation.

2. EXPERIMENTAL

2.1. Sample preparation

The powder sample of $\text{LiBaBO}_3:_{0.02}\text{Er}^{3+},_{0.08}\text{Yb}^{3+}$ was prepared by using solution combustion synthesis[^{xii}, ^{xiii}]. Several borate host materials were successfully synthesized using this method [xiv]. The stoichiometric amounts of high purity starting materials LiNO_3 ; H_3BO_3 ; NH_2CONH_2 ; NH_4NO_3 ; $\text{Ba}(\text{NO}_3)_2$ and Er_2O_3 ; Yb_2O_3 was used for phosphors preparation.

The stoichiometric amounts of the ingredients was mixed in an agate mortar with adding little amount of double distilled water. The materials then transferred into china basin. It was heated on heating menthol at about 70°C so as to obtained clear solution. The solution was then introduced into a pre-heated muffle furnace maintained at temperature 550 °C for combustion.

The solution boils and ignites to burn with flame which gave a voluminous, foamy powder. Following the combustion, the resulting foamy samples were crushed to obtain fine powder and then heated at temperature 750°C for 2 hr and suddenly cooled to room temperature.

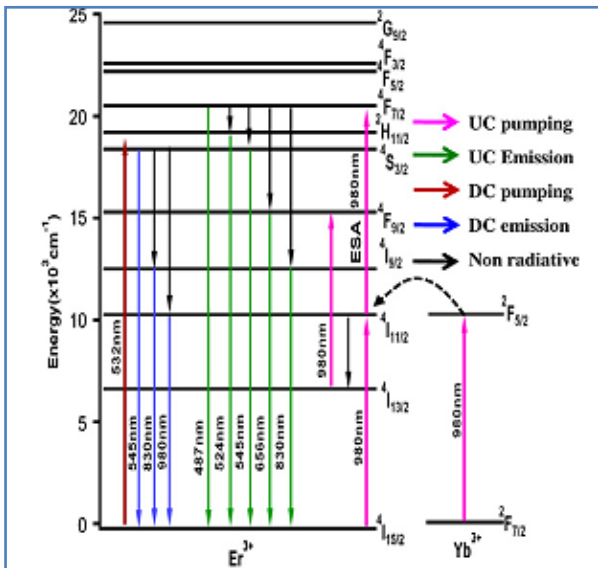
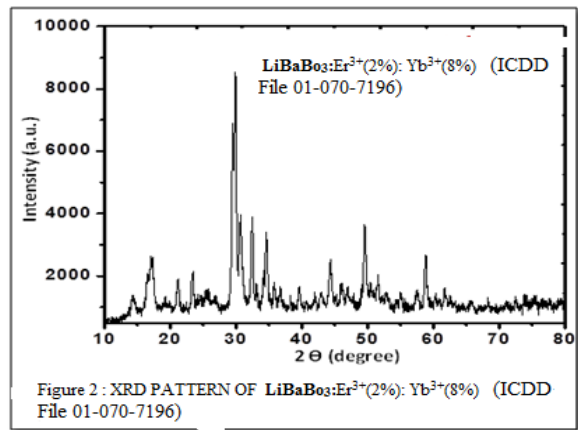


Figure (1):- Energy levels of Er³⁺ and Yb³⁺ ions showing different electronic transitions. [source:- Subrata Das, A. Amamath Reddy, G. Vijaya Prakash, " Strong green upconversion emission from Er³⁺-Yb³⁺ co-doped KCaBO₃ phosphor", Chemical Physics Letters 504 (2011) 206–210]

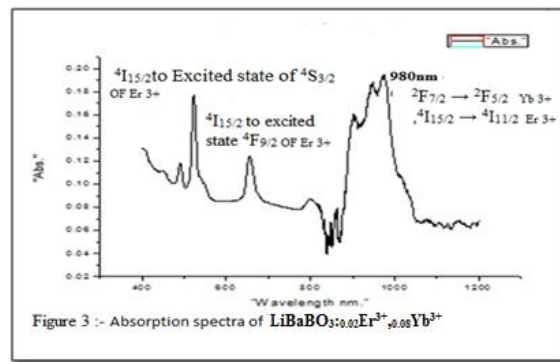


3.2. Absorption analysis of LiBaBO₃:0.02Er³⁺,0.08Yb³⁺ phosphor

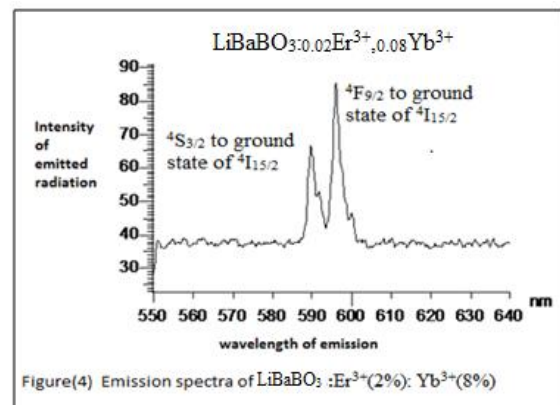
To decide excitation wavelength of sample for study of UC we did the absorption study of sample using Shimadzu UV-VIS-NIR Spectrophotometer in the range 400 nm to 1200 nm. Figure(3) shows absorption spectra of LiBaBO₃:0.02Er³⁺,0.08Yb³⁺. It shows the wide absorption band from 820 nm to 1080 nm and maximum intensity at 980 nm (Resonant transition from ⁴I_{15/2} to ⁴I_{11/2} in Er³⁺ and ²F_{7/2} to ²F_{5/2} in Yb³⁺). It also shows absorption peaks at 545 nm (from ⁴I_{15/2} to ⁴S_{3/2} transition in Er³⁺) and 656 nm (4I_{15/2} to ⁴F_{9/2} transition in Er³⁺). Due to both activator and sensitizer ions host material shows wide absorption from visible to IR.

Table (1): The molar ratio of starting materials taken for phosphor synthesis

Compound	Molar ratio and weight of chemicals (A.R.GRDE)
LiBaBO ₃ :Er ³⁺ (2 %): Yb ³⁺ (8%)	LiNO ₃ :Ba(NO ₃) ₂ :H ₃ BO ₃ :NH ₂ CoN H ₂ :NH ₄ NO ₃ :Er ₂ O ₃ :Yb ₂ O ₃
	1 : 1 : 1 : 4 : 4.5 : 0.02 : 0.08 0.6894gm:2.35255gm:0.6183:2.4029gm:3.6018gm:3.65ml:31.52ml



3.3. Photoluminescence analysis of LiBaBO₃:0.02Er³⁺,0.08Yb³⁺ phosphor



2.2. Material characterizations

The phase and surface morphology of as prepared phosphors were characterized by powder X-ray diffraction pattern using Rigaku Miniflex II X-ray Diffractometer with Cu K α radiation ($\lambda=1.54059 \text{ \AA}$) with scan speed 2 $^\circ$ /min and field emission - scanning electron microscopy (FE-SEM) (Hitachi, Model-S4800 type II). The PL measurements at room temperature were performed on Hitachi F-7000 Spectrofluorometer with spectral resolution of 2.5 nm.

3. RESULTS AND DISCUSSION

3.1. X-ray Diffraction Pattern

Fig.2 shows the powder X-ray diffraction (XRD) patterns of LiBaBO₃:0.02Er³⁺,0.08Yb³⁺, and it was found to be in good agreement with the reported standard data in ICDD file no. 01-070-7196. The result clearly implies that the obtained samples are single phase and the doping of Er³⁺ and Yb³⁺ does not cause any significant change to the detection limit of the technique in the host structure.

Photoluminescence characteristics of $\text{LiBaBO}_3:0.02\text{Er}^{3+}, 0.08\text{Yb}^{3+}$ is shown in figure(4). It shows two upconversion emission peaks at 590 nm and 596 nm when excited in infrared region. It is due to mixed transitions from $^4\text{F}_{9/2}$ and $^4\text{S}_{3/2}$ to $^4\text{I}_{15/2}$ in the Er^{3+} .

3.4 CIE chromaticity coordinates for $\text{LiBaBO}_3:0.02\text{Er}^{3+}, 0.08\text{Yb}^{3+}$ phosphor ($\lambda_{\text{ex}} = 980 \text{ nm}$)

CIE chromaticity coordinates for $\text{LiBaBO}_3:0.02\text{Er}^{3+}, 0.08\text{Yb}^{3+}$ phosphor at $\lambda_{\text{ex}} = 980 \text{ nm}$ for emission wavelength 596nm are shown in the figure (4). It shows the colour of emitted light was orange .

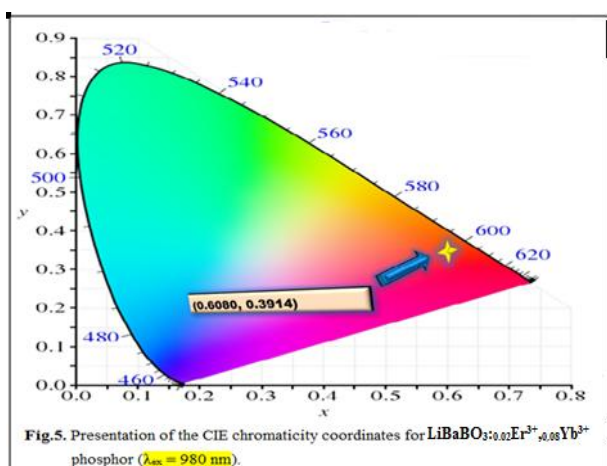


Fig.5. Presentation of the CIE chromaticity coordinates for $\text{LiBaBO}_3:0.02\text{Er}^{3+}, 0.08\text{Yb}^{3+}$ phosphor ($\lambda_{\text{ex}} = 980 \text{ nm}$).

4. CONCLUSIONS

Single phase monoclinic $\text{LiBaBO}_3:0.02\text{Er}^{3+}$ phosphors doped with 0.08Yb^{3+} concentration has been synthesized by solution combustion technique and upconversion from near-infrared to visible was reported for the first time. Their absorption spectra were recorded which shows the broad absorption range of material. Absorption band is broad in infrared region. Absorption intensity is maximum at 980 nm. The upconversion spectra of as synthesized phosphor showed Natrium Yellow and Amber (590nm and 596nm) emission of Er^{3+} , due to the mixed transitions from $^4\text{F}_{9/2}$ and $^4\text{S}_{3/2}$ to $^4\text{I}_{15/2}$ in the Er^{3+} . The intense upconversion emission from these mixed borate phosphor could be found very useful in different emerging fields.

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REFERENCES

[1] Dev K. Chatterjee, Muthu Kumara Gnanasammandhan, Yong Zhang, "Small Upconverting Fluorescent Nanoparticles for Biomedical Applications"

- [2] Parthiban Ramasamy, Palanisamy Manivasakan and Jinkwon Kim, "Upconversion nanophosphors for solar cell applications" RSC Advances issue 66, 2014.
- [3] Janet Milliez, Alexandra Rapaport, Michael Bass, Arlete Cassanho, and Hans P. Jenssen, "High-Brightness White-Light Source Based on Up-Conversion Phosphors" Journal of Display Technology
- [4] G. Bertrand-Chadeyron, R. Mahiou, M. El-Ghozzi, A. Arbus, D. Zambon, and J. C. Cousseins, "Luminescence of the orthoborate $\text{YBO}_3: \text{Eu}^{3+}$. Relationship with crystal structure," Journal of Luminescence, vol. 72–74, pp. 564–566, 1997.
- [5] P.A. Nagpure, S.K. Omanwar, "Synthesis and luminescence characteristics of terbium(III) activated NaSrBO_3 " JOURNAL OF RARE EARTHS, Vol. 30, No. 9, Sep. 2012, P. 856
- [6] N.S. Bajaj, S.K. Omanwar, "Combustion synthesis and luminescence characteristic of rare earth activated LiCaBO_3 " JOURNAL OF RARE EARTHS, Vol. 30, No. 10, Oct. 2012, P. 1005
- [7] H. Huppertz and G. Heymann, "Multianvil high-pressure/high-temperature preparation, crystal structure, and properties of the new oxoborate $\beta\text{-ZnB}_4\text{O}_7$," Solid State Sciences, vol. 5, no. 2, pp.281–289, 2003.
- [8] Xuebin Qiao a , Yu Cheng b , Lin Qin b,c , Chuanxiang Qin b , Peiqing Cai c , Sun Il Kim c , Hyo Jin Seo c,," Coprecipitation synthesis, structure and photoluminescence properties of Eu^{3+} -doped sodium barium borate" Journal of Alloys and Compounds 617 (2014) 946–951
- [9] Chr. Wyss, W. Liithy, H.P. Weber, P. Rogin, J. Hulliger, "Energy transfer in $\text{Yb}^{3+}:\text{Er}^{3+}:\text{YLF}$ " Optics Communications 1-M (19Y7) 31-35
- [10] Subrata Das, A. Amarnath Reddy, G. Vijaya Prakash, "Strong green upconversion emission from $\text{Er}^{3+}-\text{Yb}^{3+}$ co-doped KCaBO_3 phosphor", Chemical Physics Letters 504 (2011) 206–210
- [11] Sangmoon Park , So-Hye Cho, "Spectral-converting behaviors of Er^{3+} and $\text{Er}^{3+}-\text{Yb}^{3+}$ doped YOCl phosphors "Journal of Alloys and Compounds 584 (2014) 524–529
- [12] N.S. Bajaj, S.K. Omanwar, "Combustion synthesis and luminescence characteristic of rare earth activated LiCaBO_3 " JOURNAL OF RARE EARTHS, Vol. 30, No. 10, Oct. 2012, P. 1005
- [13] P.A. Nagpure, S.K. Omanwar, "Synthesis and luminescence characteristics of terbium(III) activated NaSrBO_3 " JOURNAL OF RARE EARTHS, Vol. 30, No. 9, Sep. 2012, P. 856
- [14] R. S. Palaspar, A. B. Gawande, R. P. Sonekar, S. K. Omanwar, J. Lumin. 154, 58 (2014)